

## 2,2'-Biimidazol-1-ium trichloroacetate

Ji-Xin Yuan

School of Chemistry and Materials Science,  
Wenzhou University, Wenzhou 325027,  
People's Republic of ChinaCorrespondence e-mail:  
zkrist2005@yahoo.com.cn

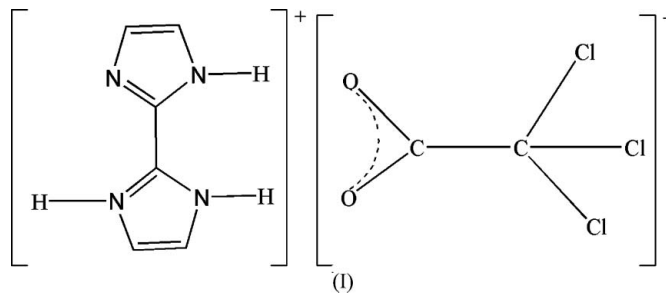
## Key indicators

Single-crystal X-ray study  
 $T = 298$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.005$  Å  
 $R$  factor = 0.055  
 $wR$  factor = 0.123  
Data-to-parameter ratio = 13.1For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

The 2,2'-biimidazol-1-ium trichloroacetate ion pairs in the title compound,  $\text{C}_6\text{H}_7\text{N}_4^+\cdot\text{C}_2\text{Cl}_3\text{O}_2^-$ , are held together by two  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bonds, and two adjacent ion pairs are linked into a centrosymmetric dimer by a pair of  $\text{N}-\text{H}\cdots\text{N}$  hydrogen bonds. The hydrogen-bonding pattern can be described in graph-set motif notation as  $R_2^2(9)$  and  $R_2^2(10)$ . Moreover, the adjacent dimers are associated by  $\pi-\pi$  interactions between five-membered rings of the 2,2'-biimidazol-1-ium cations [at  $(x, y, z)$  and  $(2-x, -y, 1-z)$ ], with the ring centroids separated by 3.861 (1) Å, forming a ribbon-like supramolecular array along the  $b$  axis.

## Comment

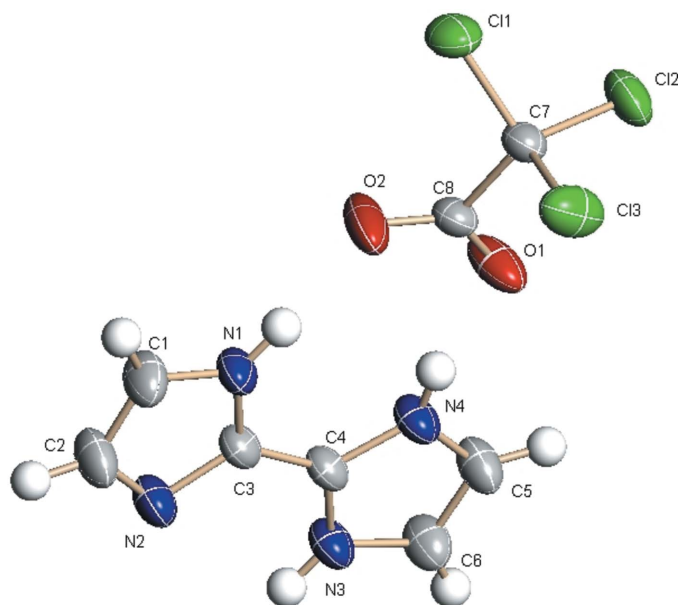
2,2'-Biimidazole,  $\text{H}_2\text{biim}$ , is not only a proton donor, but also a proton acceptor, so that it possesses five possible forms, *viz.* di-deprotonated (dianion,  $\text{biim}^{2-}$ ), mono-deprotonated (mono-anion,  $\text{Hbiim}^-$ ), neutral (neutral,  $\text{H}_2\text{biim}$ ), mono-protonated (monocation,  $\text{H}_3\text{biim}^+$ ) and di-protonated (dication,  $\text{H}_4\text{biim}^{2+}$ ). Therefore,  $\text{H}_2\text{biim}$  is an excellent candidate for the development of supramolecular motifs in crystal structures. Homomeric hydrogen-bonded motifs  $R_2^2(10)$  (Cromer *et al.*, 1987), heteromeric hydrogen-bonded motifs  $R_2^2(9)$  (Ye *et al.*, 2005) and  $R_2^1(7)$  (Bélanger & Beauchamp, 1996), and mixed hydrogen-bonded motifs  $R_2^2(10)$  and  $R_2^1(7)$  (Ramirez *et al.*, 2002), have been structurally reported. In an extension of this research, the crystal structure of the title compound,  $\text{C}_6\text{H}_7\text{N}_4^+\cdot\text{C}_2\text{Cl}_3\text{O}_2^-$ , (I), is reported here.



The bond distances and angles of the mono-protonated  $\text{H}_3\text{biim}^+$  in (I) are unexceptional and compare well with the values in neutral  $\text{H}_2\text{biim}$  (Cromer *et al.*, 1987) (Table 1 and Fig. 1). The two rings are almost coplanar in both cases. The dihedral angle between the two five-membered rings in neutral  $\text{H}_2\text{biim}$  is  $4.6^\circ$ , and is slightly smaller in (I) at  $4.47(3)^\circ$ .

Two  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bonds connect the  $\text{H}_3\text{biim}^+$  cations and trichloroacetate anions to produce ion pairs, and two adjacent ion pairs are linked into a dimer by a third  $\text{N}_3-\text{H}_3\text{A}\cdots\text{N}_2^1$  hydrogen bond and its inversion-related equiva-

Received 10 August 2005  
Accepted 13 September 2005  
Online 17 September 2005

**Figure 1**

The ion pair of (I), with the atom numbering, showing displacement ellipsoids at the 50% probability level.

lent [Table 2; symmetry code: (i)  $2 - x, 1 - y, 1 - z$ ]. The hydrogen-bonding pattern, as shown in Fig. 2, can be described in graph-set motifs (Etter, 1990; Grell *et al.*, 2000) as  $R_2^2(9)$  and  $R_2^2(10)$ . Adjacent dimers are associated by  $\pi$ - $\pi$  interactions between the five-membered rings of the  $H_3biim^+$  cations [at  $(x, y, z)$  and  $(2 - x, -y, 1 - z)$ ], with the ring centroids separated by 3.861 (1) Å, forming a ribbon-like supramolecular array along the  $b$  axis (Fig. 3).

## Experimental

2,2'-Biimidazole (2 mmol, 0.28 g) was suspended in water (30 ml). To the resulting suspension, concentrated aqueous trichloroacetic acid was added until the suspension became clear. The resulting solution was filtered and allowed to evaporate slowly at room temperature. After three weeks, colourless crystals of (I) appeared.

### Crystal data

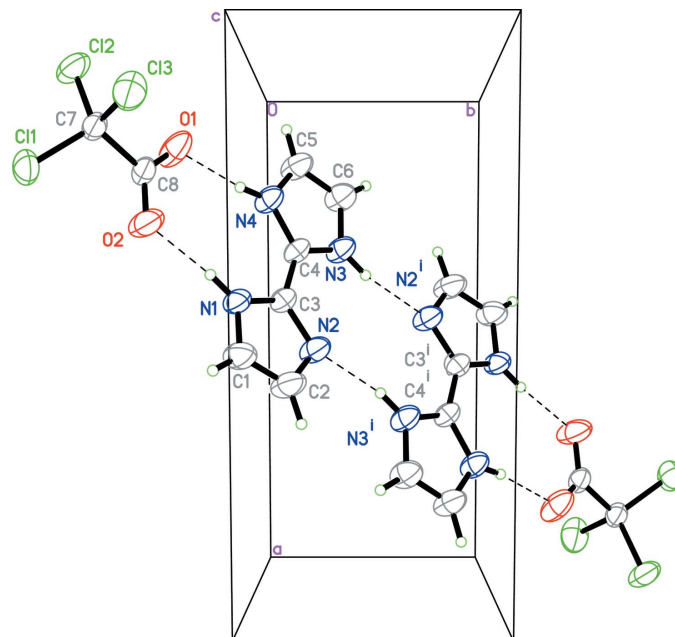
$C_6H_7N_4^+ \cdot C_2Cl_3O_2^-$   
 $M_r = 297.53$   
 Monoclinic,  $P2_1/c$   
 $a = 12.4010$  (13) Å  
 $b = 5.5664$  (6) Å  
 $c = 17.3648$  (18) Å  
 $\beta = 100.858$  (2)°  
 $V = 1177.2$  (2) Å<sup>3</sup>  
 $Z = 4$

$D_x = 1.679$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 Cell parameters from 1438 reflections  
 $\theta = 2.6$ – $24.2$ °  
 $\mu = 0.77$  mm<sup>-1</sup>  
 $T = 298$  (2) K  
 Block, colourless  
 $0.31 \times 0.13 \times 0.12$  mm

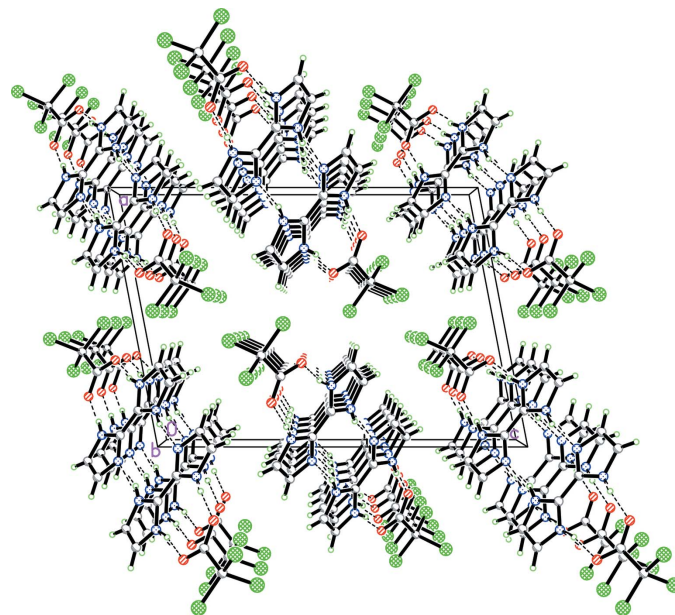
### Data collection

Bruker APEX area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Bruker, 2002)  
 $T_{\min} = 0.796$ ,  $T_{\max} = 0.913$   
 5901 measured reflections

2129 independent reflections  
 1812 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.031$   
 $\theta_{\text{max}} = 25.3$ °  
 $h = -14 \rightarrow 14$   
 $k = -6 \rightarrow 6$   
 $l = -20 \rightarrow 15$

**Figure 2**

The  $R_2^2(9)$  and  $R_2^2(10)$  hydrogen-bonding motifs in (I), formed by  $N-H \cdots O$  and  $N-H \cdots N$  hydrogen-bond interactions, which are shown as dashed lines. [Symmetry code: (i)  $2 - x, 1 - y, 1 - z$ .]

**Figure 3**

A perspective view, along the  $b$  axis, of the supramolecular array in (I).

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.055$   
 $wR(F^2) = 0.124$   
 $S = 1.14$   
 2129 reflections  
 163 parameters  
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0437P)^2 + 0.9103P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.31$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.28$  e Å<sup>-3</sup>

**Table 1**  
Selected geometric parameters (Å, °).

O1—C8	1.221 (4)	N3—C6	1.373 (4)
O2—C8	1.222 (4)	N4—C4	1.335 (4)
N1—C3	1.338 (4)	N4—C5	1.363 (4)
N1—C1	1.359 (4)	C1—C2	1.341 (5)
N2—C3	1.325 (4)	C3—C4	1.443 (4)
N2—C2	1.368 (4)	C5—C6	1.330 (5)
N3—C4	1.324 (4)	C7—C8	1.559 (4)
C3—N1—C1	107.4 (3)	N3—C4—N4	108.1 (3)
C3—N2—C2	104.7 (3)	N3—C4—C3	125.1 (3)
C4—N3—C6	108.3 (3)	N4—C4—C3	126.8 (3)
C4—N4—C5	108.6 (3)	C6—C5—N4	107.4 (3)
C2—C1—N1	106.2 (3)	C5—C6—N3	107.6 (3)
C1—C2—N2	110.5 (3)	O1—C8—O2	128.6 (3)
N2—C3—N1	111.3 (3)	O1—C8—C7	115.9 (3)
N2—C3—C4	123.9 (3)	O2—C8—C7	115.4 (3)
N1—C3—C4	124.7 (3)		

**Table 2**  
Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N1—H1A...O2	0.85 (2)	1.85 (2)	2.688 (4)	172 (3)
N3—H3A...N2 <sup>i</sup>	0.84 (2)	2.00 (2)	2.811 (4)	163 (3)
N4—H4A...O1	0.85 (2)	1.79 (2)	2.630 (4)	167 (3)

Symmetry codes: (i)  $-x + 2, -y + 1, -z + 1$ .

The H atoms of all N atoms were located in difference density maps and refined, with N—H distances restrained to 0.85 (2) Å and

with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{N})$ . The other H atoms were positioned geometrically and allowed to ride on their parent atoms at distances of  $Csp^2-H = 0.93$  Å and with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{parent atom})$ .

Data collection: *SMART* (Bruker, 2002); cell refinement: *SAINT* (Bruker, 2002); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2002); software used to prepare material for publication: *SHELXL97*.

We acknowledge financial support by the Zhejiang Provincial Technology Project Foundation of China (grant No. 2004C32088), the Zhejiang Provincial Natural Science Foundation of China (grant No. 202137) and the National Natural Science Foundation of China (grant No. 20471043).

## References

- Bélanger, S. & Beauchamp, A. L. (1996). *Acta Cryst.* **C52**, 2588–2590.
- Bruker (2002). *SADABS* (Version 2.03), *SAINT* (Version 6.02), *SMART* (Version 5.62) and *SHELXTL* (Version 6.10). Bruker AXS Inc., Madison, Winsconsin, USA.
- Cromer, D. T., Ryan, R. R. & Storm, C. B. (1987). *Acta Cryst.* **C43**, 1435–1437.
- Etter, M. C. (1990). *Acc. Chem. Res.* **23**, 120–126.
- Grell, J., Bernstein, J. & Timhofer, G. (2000). *Acta Cryst.* **B56**, 166–179.
- Ramirez, K., Reyes, J. A., Briceno, A. & Atencio, R. (2002). *CrystEngComm*, **4**, 208–212.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Ye, B. H., Ding, B. B., Weng, Y. Q. & Chen, X. M. (2005). *Cryst. Growth Des.* **5**, 801–806.